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### Separation of Prostaglandins E<sub>1</sub> and E<sub>2</sub> and other Major Eicosanoids by Unidimensional Argentation Thin Layer Chromatography

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SEPARATION OF PROSTAGLANDINS E<sub>1</sub> AND E<sub>2</sub> AND OTHER MAJOR  
ICOSANOIDS BY UNIDIMENSIONAL ARGENTATION THIN LAYER  
CHROMATOGRAPHY

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ABSTRACT

Prostaglandins (PG) E<sub>1</sub> and E<sub>2</sub> are important regulators of biologic functions, and can express different biological effects. In thin layer chromatography (TLC) systems which separate these compounds, comigration of other major eicosanoids is a problem. This paper describes a TLC system using a mobile phase of chloroform/methanol/acetic acid/H<sub>2</sub>O (90:7.5:5:0.8) that separates PGE<sub>1</sub> and PGE<sub>2</sub>, as well as other major eicosanoids, including dihomogammalinolenic acid (DHGA), the immediate fatty acid precursor of PGE<sub>1</sub>.

INTRODUCTION

PGE<sub>1</sub> and PGE<sub>2</sub> have similar biological actions, but do have different effects on some cell functions. For example, PGE<sub>1</sub> inhibits whereas PGE<sub>2</sub> does not influence or increases aggregation of human platelets (1-3). In addition, PGE<sub>1</sub> but not PGE<sub>2</sub> reduces production of the lymphokine macrophage migration inhibitory factor (4). Furthermore, PGE<sub>1</sub> and PGE<sub>2</sub> express quantitative differences in their ability to resorb bone (5). Several thin-layer chromatography (TLC) techniques have been used to separate PGE<sub>1</sub>

and PGE<sub>2</sub> (6-11). We have examined the mobile phases used and find that they do not separate the major cyclooxygenase products of arachidonic acid (AA) from PGE<sub>1</sub> and PGE<sub>2</sub>. This is a potential source of error if the major eicosanoids are to be quantified by bioassay or examined by scintillation counting or autoradiography. We therefore developed a simple TLC technique that separates PGE<sub>1</sub>, PGE<sub>2</sub> and thromboxane B<sub>2</sub> (TxB<sub>2</sub>), PGD<sub>2</sub>, 6-Keto-PGF<sub>1a</sub> (stable metabolite of prostacycline), and the immediate precursor of the 1 series eicosanoids, dihomogammalinolenic acid (DHLA).

#### MATERIALS AND METHODS

All chemicals were of high performance liquid chromatographic grade (Fisher Chemicals, King of Prussia, PA). Glass chromatography tanks (28 cm x 27 cm x 8 cm) with glass lids (Scientific Manufacturing Industries, Emeryville, CA) were used. Whatman LK5 silica-coated TLC plates (Whatman Chemical, Clifton, NJ) were scored into 2 cm channels with a 1 cm unused channel on each side. Plates were washed overnight in a mobile phase of chloroform/methanol (90:5), and air dried. They were coated with a 10% solution of silver nitrate (Kodak Chemicals, Rochester, NY) in distilled water and ethyl alcohol (14:86, v/v) in a 10" x 14" metal pan, air dried, activated in an oven (Boekel, Philadelphia, PA) at 110°F for 30 min, and then cooled at room air. Plates wrapped in aluminum foil and an air tight plastic bag could be kept for at least 1 week before use. All unlabeled eicosanoids were purchased from Sigma Chemical Co. (St. Louis, MO). Eicosanoids (5-10 ug) were spotted in the middle of the preadsorbant zone with a 20 ul (Rainen Instrument Co, Woburn, MA) or a 50 ul pipette (Hamilton Co., Reno, NV). After spotting, the eicosanoids were allowed to air dry, and were then run in the appropriate mobile phase until the solvent zone had reached 1 cm from the top of the plate. The plates were air dried, and then

dried at 110°F for 10 minutes. They were cooled at room air, and then sprayed with 3% cupric acetate (3 gm) in phosphoric acid/water (8:92, v/v) and air dried. Charring at 180°F for 10 minutes was often helpful in improving resolution of the bands.

### RESULTS

Various solvent systems that have been used to separate PGE<sub>1</sub> and PGE<sub>2</sub> are listed in Table 1.

Although the TLC systems referenced in Table 1 separate PGE<sub>1</sub> from PGE<sub>2</sub>, other major eicosanoids comigrate thus diminishing the utility of these systems (Table 2). However, separation of PGE<sub>1</sub>,

TABLE 1

Solvent Systems Used to Separate Eicosanoids

<u>TLC System</u>	<u>Ref.</u>	<u>Solvent Composition by Volume</u>
SI	Present paper	chloroform/methanol/acetic acid/H <sub>2</sub> O (90:7.5:5:0.8)
SII	12	ethyl acetate/2,2,4-trimethylpentane/acetic acid/methanol/H <sub>2</sub> O (110:10:30:35:100)
SIII	6	ethyl acetate/2,2,4-trimethylpentane/acetic acid/H <sub>2</sub> O (100:30:10:100)
SIV	6	ethyl acetate/2,2,4-trimethylpentane/acetic acid/H <sub>2</sub> O (90:50:20:100)
SV	13	chloroform/methanol/acetic acid (90:5:5)
SVI	14	chloroform/methanol/acetic acid/H <sub>2</sub> O (95:5:1:0.2)
SVII	9	0.0025 M phosphoric acid/acetonitrile with 0.2 M NaCl (52:48)
SVIII	8	organic (upper phase) of ethyl acetate/2,2,4-trimethylpentane/acetic acid/H <sub>2</sub> O (110:50:20:100)
SIX	8	chloroform/methanol/acetic acid (80:10:10)
SX	15	ethyl acetate/isooctane/ethanol/acetic acid/H <sub>2</sub> O (35:10:3:0.1:0.1)
SXI	11	ethyl acetate/methanol/acetic acid (100:10:1)
SXII	11	chloroform/methanol/acetic acid/H <sub>2</sub> O (90:9:1:0.65)
SXIII	11	benzene/1,4-dioxane/acetic acid (20:20:1)
SXIV	11	ethyl acetate/acetone/acetic acid (90:10:1)

TABLE 2

Mobile Phases Used to Separate PGE<sub>1</sub> and PGE<sub>2</sub> But With Known Comigration of Major Eicosanoids<sup>1</sup>

<u>TLC System</u>	<u>Ref.</u>	<u>Comigrating Eicosanoids</u>
SII	12	TxB <sub>2</sub> and PGD <sub>2</sub>
SIII	6	multiple tailing
SIV	6,11	PGE <sub>1</sub> and PGE <sub>2</sub> blur
SV	13	TxB <sub>2</sub> and PGD <sub>2</sub>
SVI	14	6-Keto-PGF <sub>1a</sub> and PGE <sub>2</sub>
SVII	9	PGE <sub>1</sub> and PGD <sub>2</sub>
SVIII	8	TxB <sub>2</sub> and PGD <sub>2</sub>
SIX	8	TxB <sub>2</sub> and 6-Keto-PGF <sub>1a</sub>
SX	15	6-Keto-PGF <sub>1a</sub> tails
SXI	11	6-Keto-PGF <sub>1a</sub> and PGF <sub>2a</sub> , PGE <sub>2</sub> and PGD <sub>2</sub>
SXII	11	PGE <sub>2</sub> and 6-Keto-PGF <sub>1a</sub>
SXIII	11	6-Keto-PGF <sub>1a</sub> and PGF <sub>2a</sub>
SXIV	11	AA tailing

<sup>1</sup> Comigration established as noted in References or by argentation technique as described in Materials and Methods.

PGE<sub>2</sub> and TxB<sub>2</sub>, PGD<sub>2</sub>, 6-Keto-F<sub>1a</sub>, PGF<sub>2a</sub>, AA and DHLA was achieved with chloroform/ methanol/acetic acid/H<sub>2</sub>O (90:7.5:5:0.8) (Solvent I) as shown in Fig. 1. The mobilities of the compounds, expressed as R<sub>f</sub> values, are shown in Table 3 with the preabsorbant-silica gel boundary as the origin. Development time is approximately 60 minutes. The method is reproducible (R<sub>f</sub> standard error of the mean .5%) and reliable.

## DISCUSSION

The method presented here allows separation by TLC of the monoenoic and dienoic PGE compounds and their immediate fatty acid precursors DHLA and AA respectively. Thus, metabolism of both PGE precursors may be studied simultaneously. In addition, the same system accommodates separation of the major cyclooxygenase products of AA: TxB<sub>2</sub>, 6-Keto-F<sub>1a</sub>, PGD<sub>2</sub>, PGF<sub>2a</sub>. Other investigators have separated PGE<sub>1</sub> from PGE<sub>2</sub>. However, as shown here, comigration of other major eicosanoids with PGE<sub>1</sub> and PGE<sub>2</sub> occurs with several of the solvent systems used. Use of the Solvent I, a mobile phase of chloroform, methanol, acetic acid and water allows separation of PGE<sub>1</sub> from PGE<sub>2</sub> as well as separation of other major eicosanoids. This system is not only useful in

TABLE 3  
Mobility of Separated Eicosanoids

<u>Compound</u>	<u>Mobility (expressed as <math>R_f</math>)</u>
PGF <sub>2a</sub>	0.12
TxB <sub>2</sub>	0.28
PGD <sub>2</sub>	0.32
PGE <sub>2</sub>	0.38
6-Keto-PGF <sub>1a</sub>	0.51
AA	0.58
PGE <sub>1</sub>	0.62
DHLA	0.83

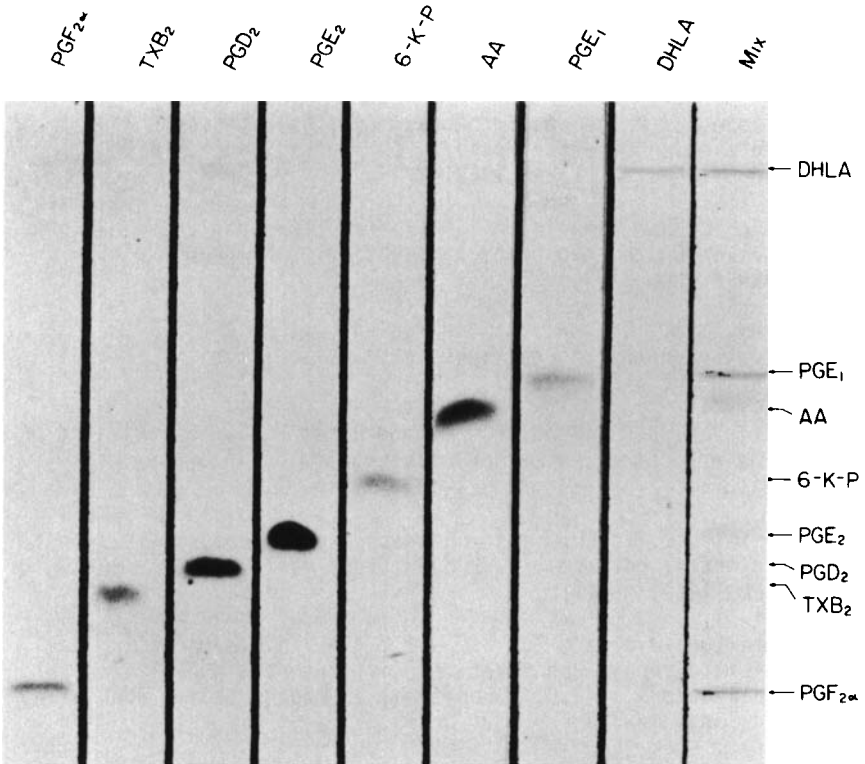


FIGURE 1

Chromatogram showing eicosanoid separation with a mobile phase of chloroform/methanol/acetic acid/H<sub>2</sub>O (90:7.5:5.0:0.8). 6-Keto-PGF<sub>1a</sub> is abbreviated as 6-K-P.

examining AA and its metabolites, but may also be used in experiments where DHLA and PGE<sub>1</sub> are to be examined.

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